

BRIEF SURVEY OF UNCERTAINTY IN PHYSICS LABS

First Step **VERIFYING THE VALIDITY OF RECORDED DATA**

The drawing of graphs during lab measurements is practical way to estimate quickly:

- a) Whether the measurements confirm the expected behaviour predicted by physics model.
- b) If any of recorded data is measured in wrong way and must be excluded from further data treatments.

Example_1: We drop an object from a window and we expect it to hit ground after 2sec. To verify our expectation, we *measure* this *time* several times and record the following results;

1.99s, 2.01s, 1.89s, 2.05s 1.96s, 1.99s, 2.68s, 1.97s, 2.03s, 1.95s

(Note: **3-5 measurements** is a **minimum acceptable number of data** for estimating a parameter, i.e. repeat the measurement 3-5 times. The estimation based on 1 or 2 data is not reliable.)

To check out those data we include them in a graph (fig.1). From this graph we can see that:

- a) The fall time seems to be *constant* and very likely ~2s. So, in general, we have acceptable data.
- b) Only the seventh measure is too far from the others results and this may be due to an abnormal circumstance during its measurement. To eliminate any doubt, we **exclude** this value from the following data analysis. We have enough other data to work with. Our remaining data are: 1.99s, 2.01s, 1.89s, 2.05s, 1.96s, 1.99s, 1.97s, 2.03s, 1.95s. .

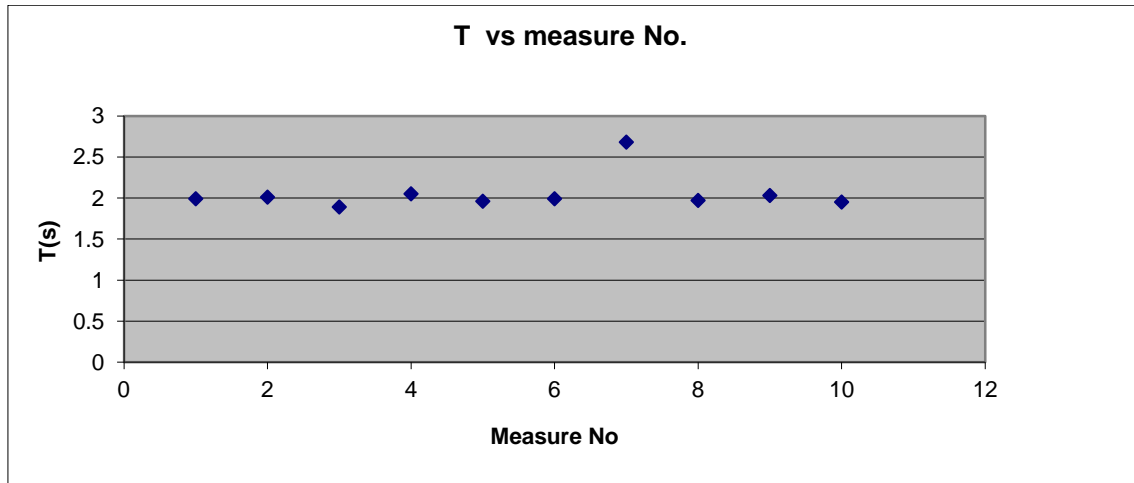


Fig.1

Second step **ORGANIZING RECORDED DATA IN A TABLE**

Include all data in a table organized in such a way that some cells be ready to include the uncertainty calculation results. In our example, we are looking to estimate a single parameter “**T**”, so we have to predict (*at least*) two cells for its average and its uncertainty.

Table_1

T ₁	T ₂	T ₃	T ₄	T ₅	T ₆	T ₇	T ₈	T ₉	T _{av}	ΔT
1.99s	2.01s	1.89s	2.05s	1.96s	1.99s	1.97s	2.03s	1.95s		

The **true value** of parameter is unknown. We use the *recorded data* to find an **estimation** of the **true value** and the **uncertainty** of this **estimation**.

There are three particular situations for uncertainty of estimations.

A] - We measure several times a parameter and we get always the same numerical value.

Example_2: We measure the length of a table three times and we get $L = 85\text{cm}$ and *a little bit more or less*. This happens because the smallest unit of the meter stick is **1cm** and we **cannot be precise** about what *portion of 1cm* is the quantity “*a little bit more or less*”. In such situations we use “**the half-scale rule**” i.e.; **the uncertainty is equal to the half of the smallest unit available used for measurement.** In our example $\Delta L = \pm 0.5\text{cm}$ and *the result of measurement is reported as* $L = (85.0 \pm 0.5)\text{cm}$.

-If we use a meter stick with **smallest unit available 1mm**, we are going to have a more precise result but even in this case there is an uncertainty. Suppose that we get always the length $L = 853\text{mm}$. Being aware that there is always a **parallax error** (eye position) on both sides reading, one may get $\Delta L = \pm 0.5, \pm 1$ and even $\pm 2\text{mm}$) depending on the measurement circumstances. **The result of measurement is reported as** $L = (853.0 \pm 0.5)\text{mm}$ or $(853 \pm 1)\text{mm}$ or $(853 \pm 2)\text{mm}$. Our **best estimation** for the table length is 853mm . Also, our measurements show that the **true length** is between 852 and 854mm. If the **absolute uncertainty** of estimation is $\Delta L = \pm 1\text{mm}$, then the **uncertainty interval** is $(852, 854)\text{mm}$.

-Let's suppose that using the same meter stick, we measure the length of a calculator and a room and find $L_{\text{calc}} = (14.0 \pm 0.5)\text{cm}$ and $L_{\text{room}} = (525.0 \pm 0.5)\text{cm}$. In the two cases we have the same absolute uncertainty $\Delta L = \pm 0.5\text{cm}$ but we are *conscious that the length of room is measured more precisely*. **The precision of a measurement is estimated by the uncertainty portion that belongs to the unit of measurement quantity. Actually, it is estimated by the relative error**

$$\varepsilon = \frac{\Delta L}{L} * 100\% \quad (1)$$

-Note that **smaller relative error** means **higher precision** of measurement. In our length measurement, we have $\varepsilon_{\text{calc}} = \frac{0.5}{14} * 100\% = 3.57\%$ and $\varepsilon_{\text{room}} = \frac{0.5}{525} * 100\% = 0.095\%$. We see that the room length is measured much more precisely (about 38 times).

Note: Don't mix the **precision** with **accuracy**. A measurement is **accurate** if **uncertainty interval** contains an **expected (known by literature) value** and **non accurate** if it does not contain it.

B] We measure several times a parameter and we get always different numerical values.

In this case, one takes **average** as the **best estimation** and **mean deviation** as **absolute uncertainty**.

Example: For data collected in *experiment_1*

b.1) The best estimation for falling time is the **average of measured data**.

$$\bar{T} = \frac{1}{n} \sum_{i=1}^n T_i = \frac{1}{9} \sum_{i=1}^9 T_i = \frac{1}{9} [1.99 + 2.01 + 1.89 + 2.05 + 1.96 + 1.99 + 1.97 + 2.03 + 1.95] = 1.982\text{s} \quad (2)$$

b.2) One uses the **spread of measured data** to get an **estimation for absolute uncertainty**.

A first way to estimate the spread is by use of **mean deviation** i.e. “*average distance*” of **data from their average value**. In the case of our example

we get

$$\Delta T = \frac{1}{n} \sum_{i=1}^n |T_i - \bar{T}| = \frac{1}{9} \sum_{i=1}^9 |T_i - 1.982| = 0.0353\text{s} \quad (3)$$

Now we can say that the **true value** of fall time is inside the **uncertainty interval** (1.947, 2.017)sec or between $T_{\max} = 2.017s$ and $T_{\min} = 1.947s$ with **best estimation** 1.982s. Taking in account the rules on *significant figures* and *rounding off* we get $T_{\text{Best}} = 1.98\text{sec}$ and $\Delta T = 0.04\text{sec}$ and

$$\text{The result is reported as } T = (1.98 \pm 0.04)\text{sec} \quad (4)$$

Another (statistically better) estimation of spread is the “**standard¹ deviation**” of data.

Based on our example data we get
$$\sigma T = \sqrt{\frac{\sum_{i=1}^n (T_i - \bar{T})^2}{n-1}} = \sqrt{\frac{\sum_{i=1}^9 (T_i - 1.982)^2}{8}} = 0.047s. \quad (5)$$

$$\text{The result is reported as } T = (1.98 \pm 0.05)\text{sec} \quad (6)$$

b.3) For *spread estimation*, a **larger interval of uncertainty** means a more “**conservative estimation**” but in the same time a **more reliable estimation**. That’s why the **standard deviation** is a better estimation for the **absolute uncertainty**. Note that we get $\Delta T = \pm 0.05s$ when using the **standard deviation** and $\Delta T = \pm 0.04s$ when using the **mean deviation**. Also, the **relative error (or relative uncertainty)** calculated from the **standard deviation** is bigger. In our example the **relative uncertainty** of measurements is

$$\varepsilon = \frac{\sigma T}{\bar{T}} * 100\% = \frac{0.047}{1.982} * 100\% = 2.37\% \quad \text{when using the } \textit{standard deviation}$$

$$\text{and } \varepsilon = \frac{\Delta T}{\bar{T}} * 100\% = \frac{0.035}{1.982} * 100\% = 1.81\% \quad \text{when using the } \textit{mean deviation}$$

Important: The absolute and relative uncertainty can never be zero.

Assume that you **repeat** 5 times a given measurement and you read all times the same value X . So, by applying the rules of case “b” you may report $X_{\text{best}} = X_{\text{Av}} = 5X/5 = X$ and $\Delta X_b = 0$. **But here you deal with a case “a”** and this means that there is a $\Delta X_a (\neq 0) = 1/2(\text{smallest unit of measurement scale})$. This example shows that, when calculating the absolute uncertainty, *one should take into account the precise expression*

$$\Delta X = \Delta X_a + \Delta X_b \quad (7)$$

Note that in those cases where $\Delta X_b \gg \Delta X_a$ *one may simply disregard ΔX_a* .

Example: In example_1 the time is measured with 2 decimals. This means that $\Delta X_a = 1/2(0.01) = 0.005s$ Meanwhile (from 6) $\Delta X_b = 0.05s$ which is **ten times bigger** than ΔX_a . *In this case one may neglect ΔX_a . But if ΔX_b were $0.02s$ and $\Delta X_a = 0.005s$ one cannot neglect $\Delta X_a = 0.005s$ because it is 25% of ΔX_b . In this case one must use the expression (7) to calculate the absolute uncertainty and $\Delta X = 0.02 + 0.005 = 0.025s$*

Note: You will consider that a measurement has a *good precision* if the **relative uncertainty $\varepsilon < 10\%$** .

If the relative uncertainty is $\varepsilon > 10\%$, you may proceed by:

- Cancelling* any particular data “**shifted too much from the average value**” ;
- Increasing* the number of data by repeating more times the measurement;
- Improving* the measurement procedure.

¹ The standard deviation can be calculated direct in Excel and in many calculators.

C] Estimation of Uncertainties for Calculated Quantities (Uncertainty propagation)

Very often, we use the experimental data recorded for some parameters and a mathematical expression to estimate the value of a given parameter of interests (POI). As we estimate the *measured parameters* with a certain *uncertainty*, it is clear that the estimation of POI will have some uncertainty, too.

Actually, the calculation of *best estimation* for POI is based on the *best estimations of measured parameters* and the formula that relates POI with measured parameters. Meanwhile, the uncertainty of POI estimation is calculated by using the *Max_Min* method. This method calculates the limits of uncertainty interval, *POI_{min}* and *POI_{max}* by using the formula relating POI with other parameters and the combination of their limiting values in such a way that the result be the smallest or the largest possible.

Example. To find the volume of a rectangular pool with constant depth, we measure its length L, its width W and its depth D by a meter stick. Then, we calculate the volume by using the formula $V=L*W*D$. Assume that our measurement results are $L = (25.5 \pm 0.5)\text{m}$, $W = 12.0 \pm 0.5\text{m}$, $D = 3.5 \pm 0.5\text{m}$

In this case the **best estimation** for the volume is $V_{\text{best}} = 25.5*12.0*3.5=1071.0 \text{ m}^3$. This estimation of volume is associated by an uncertainty calculated by *Max-Min methods* as follows

$$V_{\text{min}}=L_{\text{min}}*W_{\text{min}}*D_{\text{min}}= 25*11.5*3 = 862.5\text{m}^3 \quad \text{and} \quad V_{\text{max}}=L_{\text{max}}*W_{\text{max}}*D_{\text{max}}= 26*12.5*4 = 1300.0\text{m}^3$$

So, the *uncertainty interval* for volume is (862.5, 1300.0) and the *absolute uncertainty* is

$$\Delta V = (V_{\text{max}}-V_{\text{min}})/2 = (1300.0 - 862.5)/2 = 218.7\text{m}^3 \quad \text{while the relative error is } \varepsilon_V = \frac{218.7}{1071.0} * 100\% = 20.42\%$$

Note_1: When applying the Max-Min method to calculate the uncertainty, one must pay attention to the mathematical expression that relates POI to measured parameters.

Examples: - You *measure* the *period* of an oscillation and you use it to *calculate* the *frequency* (POI).
As $f = 1/T$, $f_{\text{av}} = 1/T_{\text{av}}$ the *max-min method* gives $f_{\text{min}}=1/T_{\text{max}}$ and $f_{\text{max}}=1/T_{\text{min}}$
- If $z = x - y$, $z_{\text{av}} = x_{\text{av}} - y_{\text{av}}$ and $z_{\text{MAX}} = x_{\text{MAX}} - y_{\text{MIN}}$ and $z_{\text{MIN}} = x_{\text{MIN}} - y_{\text{MAX}}$.

Note_2. Use the *best estimations of parameters in the expression* to calculate the *best estimation for POI*. If they are missing one may use POI_{middle} as the best estimation for POI

$$POI_{\text{middle}} = \frac{POI_{\text{MAX}} + POI_{\text{MIN}}}{2} \quad (8)$$

Be aware though, that POI_{middle} is not always equal to **POI best estimation**.

So, for the pool volume $V_{\text{middle}} = (1300+862.5)/2 = 1081.25\text{m}^3$ which is different from $V_{\text{best}} = 1071.0 \text{ m}^3$

How to present the result of uncertainty calculations? You must provide the **best estimation**, the **absolute uncertainty** and the **relative uncertainty**. So, for the last example, the result of uncertainty calculations should be presented as follows: $V = (1071.0 \pm 218.7) \text{ m}^3$, $\varepsilon = (218/1071)*100\% = 20.42\%$

Note: **Absolute uncertainties** must be *quoted* to the **same number of decimals** as the **best estimation**. The use of *scientific notation* helps to prevent confusion about the number of significant figures.

Example: If calculations generate, say $A = (0.03456789 \pm 0.00245678)\text{m}$
This should be presented after being rounded off (leave 1,2 or 3 digits after decimal point):

$$A = (3.5 \pm 0.2) * 10^{-2}\text{m} \quad \text{or} \quad A = (3.46 \pm 0.25) * 10^{-2}\text{m}$$

HOW TO CHECK WHETHER TWO QUANTITIES ARE EQUAL?

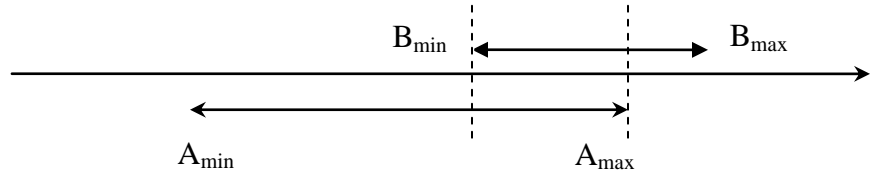
This question appears essentially in two situations:

1. We measure the *same parameter* by two *different methods* and want to verify if the results are equal.
2. We use measurements to *verify* if a *theoretical expression* is right.

In the first case, we have to compare the estimations $A \pm \Delta A$ and $B \pm \Delta B$ of the “two parameters”. The second case can be transformed easily to the first case by noting the left side of expression A and the right side of expression B . Then, the procedure is the same. **Example:** We want to verify if the thin lens equation $1/p + 1/q = 1/f$ is right. For this we note $1/p + 1/q = A$ and $1/f = B$

Rule: We will consider that the quantities A and B are *equal*² if their *uncertainty intervals overlap*.

Fig.2



WORK WITH GRAPHS

We use graphs to *check the theoretical expressions* or to *find the values* of physical quantities.

Example; We find theoretically that the oscillation period of a *simple pendulum* is $T = 2\pi\sqrt{L/g}$ and we want to verify it experimentally. For this, as a first step, we prefer to get a linear relationship between two quantities we can measure; in our case period T and length L . For this we square the two sides of the

relation $T^2 = \frac{4\pi^2}{g} * L$ pose $T^2 = y, L = x$ and get the linear expression $y = a*x$ where $a = 4\pi^2/g$.

So, we have to verify experimentally if there is such a relation between T^2 and L . *Note that if this is verified we can use the experimental value of “a” to calculate the free fall constant value “g = 4π²/a”.*

- Assume that after measuring the period for a given pendulum length several times, calculated the *average values* and *uncertainties* for $y(=T^2)$ and repeated this for a set of different values of length $x(L=1, \dots, 6m)$, we get the data shown in table No 1. At first, we graph the average data (X_{av}, Y_{av}) . We see that they are aligned on a straight line, as expected. Then, we use Excel to find the best linear fitting for our data and we ask this line to pass from $(x = 0, y = 0)$ because *this is predicted from the theoretical formula*. We get a straight line with $a_{av} = 4.065$. Using our theoretical formula we calculate the estimation for $g_{av} = 4\pi^2/a_{av} = 4\pi^2/4.065 = 9.70$ which is not far from *expected value 9.8*. Next, we add the *uncertainties in the graph* and draw the best linear fitting with *maximum /minimum slope* that pass by origin. From *the graphs* we get $a_{min} = 3.635 / a_{max} = 4.202$. So, $g_{min} = 4\pi^2/a_{max} = 4\pi^2/4.202 = 9.38$ and $g_{max} = 4\pi^2/a_{min} = 4\pi^2/3.635 = 10.85$

Table_2

X	Y(av.)	ΔY (+/-)	Ymin	Ymax	Max. Slope	Min. Slope
1	4	1.5	2.5	5.5	4.202	3.635
2	8.3	1.8	6.5	10.1		
3	11.8	1.3	10.5	13.1	P1 (1; 1.5)	P1 (1; 5.5)
4	17	1.6	15.4	18.6	P2 (6; 25.5)	P2 (6; 21.5)
5	21	1.1	19.9	22.1		
6	23.5	2	21.5	25.5		

This way, by using the graphs we:

- 1- have *proved experimentally* that *our relation* between T and L is right.
- 2- find out that our measurements are *accurate* because the *uncertainty interval* (9.38, 10.85) for “g” does include the *officially accepted value* $g = 9.8m/s^2$
- 3- find the *absolute error* $\Delta g = (10.85 - 9.38)/2 = 0.735m/s^2$
The relative error is $\epsilon = (0.735/9.70) * 100\% = 7.6\%$ which means a acceptable ($< 10\%$) *precision of measurement*.

² They should be expressed in the same unit, for sure.

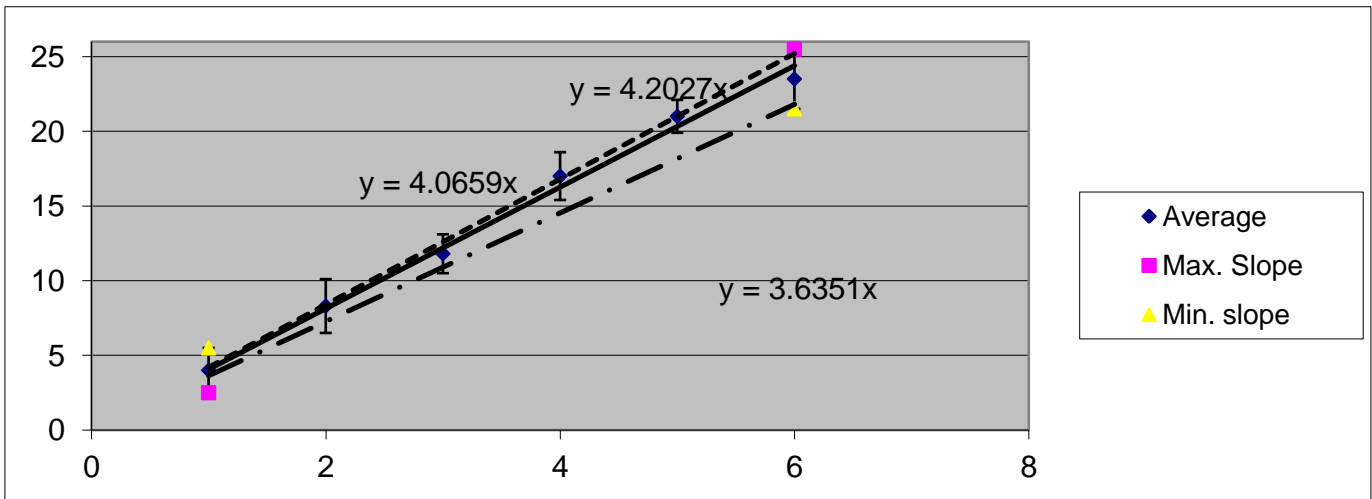
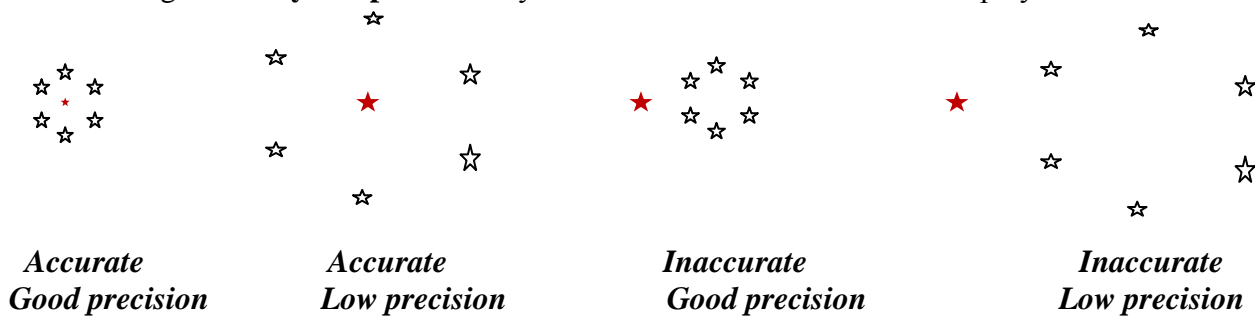


Fig.3

ABOUT THE ACCURACY AND PRECISION

- Understanding **accuracy** and **precision** by use of hits distribution in a Dart's play.



- As a rule, before using a method (or device) for measurements, one should verify that the method produces **accurate results** in the range of expected values for the parameter under study. This is an obligatory step in research and industry and it is widely known as the **calibration procedure**. During a calibration procedure one records a set of data and makes sure that the **result is accurate**.

In principle, the *result of experiment is accurate* if the “average of data” fits to the “officially accepted value”. We will consider that our experiment is “enough accurate” if the “officially accepted value” falls **inside** the interval of uncertainty of measured parameter; otherwise the result is inaccurate.

The quantity $\varepsilon_{\text{accu}} = \frac{|C_{\text{Av}} - C_{\text{off}}|}{C_{\text{off}}} \times 100 \%$ (often ambiguously named as **error**) gives the **relative shift of**

average from the officially accepted value C_{official} . It is clear that **the accuracy is higher** when $\varepsilon_{\text{accu}}$ is smaller. **But, the measurement is inaccurate if $\varepsilon_{\text{accu}} > \varepsilon$ (relative uncertainty of measurement).**

Remember that relative uncertainty $\varepsilon = \frac{\Delta C}{C_{\text{Av}}} \times 100 \%$ **is different from** $\varepsilon_{\text{accu}}$.

Note: For an **a big number of measured data** and **accurate measurement**, the average should fit to the expected value of parameter and $\varepsilon_{\text{accu}}$ **should be practically zero**. Meanwhile the **relative error** ε **tends to a fixed value different from zero**. Actually, ε can never be equal to zero.